

2,4,6-Tris(4-pyridylmethylsulfanyl)-1,3,5-triazine  
monohydrateJarrod J. M. Amoore and  
Cameron J. Kepert\*School of Chemistry, F11, University of Sydney,  
NSW 2006, AustraliaCorrespondence e-mail:  
c.kepert@chem.usyd.edu.au

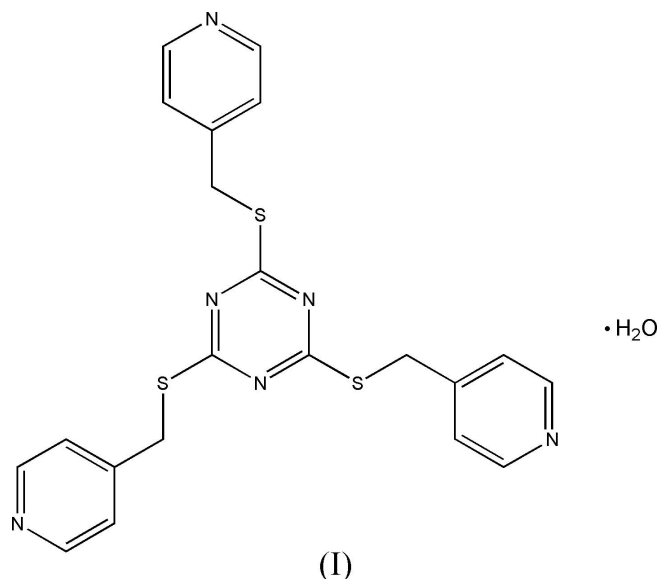
## Key indicators

Single-crystal X-ray study  
 $T = 150$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.032  
 $wR$  factor = 0.086  
Data-to-parameter ratio = 17.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

In the crystal structure of the title compound,  $\text{C}_{21}\text{H}_{18}\text{N}_6\text{S}_3 \cdot \text{H}_2\text{O}$ , the ligands are held together through  $\pi$ - $\pi$  stacking interactions and  $\text{O}-\text{H} \cdots \text{N}$  hydrogen bonds involving the water molecule.

## Comment

The ligand 2,4,6-tris(4-pyridylmethylsulfanyl)-1,3,5-triazine (4-TPST) has been used previously in the formation of supramolecular structures containing silver and nickel (Hong *et al.*, 2000a). However, the ligand has not previously been crystallized without metal ions in the structure. The asymmetric unit of the title compound, (I), contains one 4-TPST molecule and one water molecule (Fig. 1). The 4-TPST molecule is arranged with two of the pyridine rings bent above the plane of the central triazine ring and the other bent down below the plane. The three  $\text{C}_{\text{triazine}}-\text{S}-\text{C}_{\text{ethyl}}$  angles [102.09 (7)–103.35 (7)°] are all arranged in the same direction around the central triazine ring, as seen in one of the literature examples (Hong *et al.*, 2000b) but not in the other (Hong *et al.*, 2000a). The central triazine rings are involved in weak  $\pi$ - $\pi$  interactions, the centroid-centroid distance being 3.9981 (5) Å, in the direction of the  $a$  axis (Janiak, 2000). There are shorter  $\pi$ - $\pi$  interactions, the centroid-centroid distance being 3.5617 (4) Å between the pyridine rings of the 4-TPST ligands, linking the ligands together in the [101] direction. There are also numerous hydrogen-bonding interactions between each water molecule and the three surrounding ligand molecules (Desiraju, 2002).

Received 9 May 2005  
Accepted 20 May 2005  
Online 28 May 2005

## Experimental

The ligand 2,4,6-tris(4-pyridylmethylsulfanyl)-1,3,5-triazine was prepared *via* literature methods (Hong *et al.*, 2000*b*). Single crystals suitable for X-ray analysis were grown by slow evaporation in air of a 1:1 solution of nitromethane and chloroform.

### Crystal data

$C_{21}H_{18}N_6S_3 \cdot H_2O$	$Z = 2$
$M_r = 468.61$	$D_x = 1.465 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.4360$ (10) Å	Cell parameters from 5238 reflections
$b = 10.8759$ (12) Å	$\theta = 2.3\text{--}28.0^\circ$
$c = 12.1904$ (14) Å	$\mu = 0.38 \text{ mm}^{-1}$
$\alpha = 73.9052$ (18)°	$T = 150$ (2) K
$\beta = 85.9455$ (19)°	Needle, light yellow
$\gamma = 81.3899$ (19)°	$0.55 \times 0.11 \times 0.09 \text{ mm}$
$V = 1062.0$ (4) Å <sup>3</sup>	

### Data collection

Bruker SMART 1000 CCD diffractometer	4892 independent reflections
$\omega$ scans	4257 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.020$
$T_{\text{min}} = 0.89$ , $T_{\text{max}} = 0.97$	$\theta_{\text{max}} = 28.1^\circ$
10553 measured reflections	$h = -10 \rightarrow 11$
	$k = -13 \rightarrow 14$
	$l = -16 \rightarrow 16$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0371P)^2 + 0.6826P]$
$R[F^2 > 2\sigma(F^2)] = 0.033$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.086$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.42 \text{ e \AA}^{-3}$
4892 reflections	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$
286 parameters	
H atoms treated by a mixture of independent and constrained refinement	

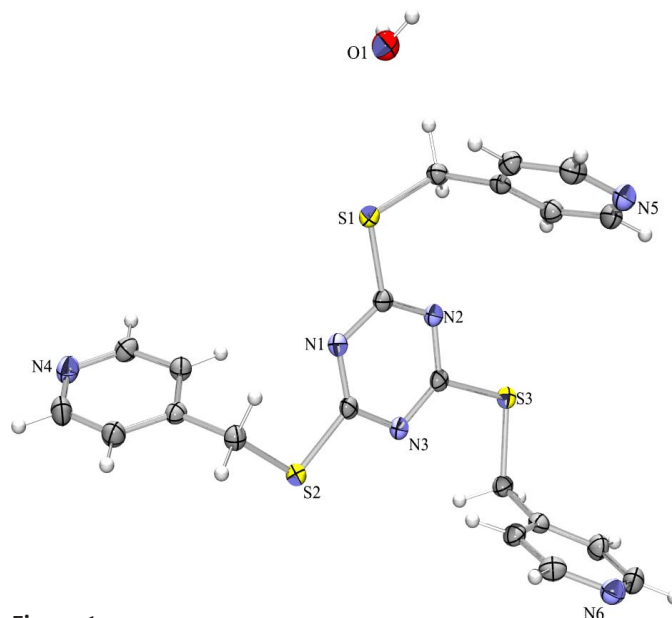
**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O1-H1A \cdots N4^i$	0.83 (2)	2.16 (2)	2.944 (2)	157 (2)
$O1-H1B \cdots N6^{ii}$	0.84 (2)	2.05 (2)	2.872 (2)	165 (2)

Symmetry codes: (i)  $x, y, z - 1$ ; (ii)  $x + 1, y - 1, z$ .

All of the H atoms of the 4-TPST molecule were placed in geometrically idealized positions and treated as riding, with  $C-H = 0.95\text{--}0.99$  Å and  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ . The H atoms on the water molecule were located in a difference Fourier map and refined with fixed isotropic displacement parameters of  $1.2U_{\text{eq}}(O)$ .



**Figure 1**  
View of the asymmetric unit. Displacement ellipsoids are drawn at the 50% probability level.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT-Plus (Bruker, 1997); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: XCIF (Bruker, 2001).

We thank the Australian Research Council for funding.

## References

- Bruker (1997). SMART and SAINT-Plus. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2001). XCIF. Bruker AXS Inc., Madison, Wisconsin, USA.
- Desiraju, G. R. (2002). *Acc. Chem. Res.* **35**, 565–573.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Hong, M., Zhao, Y., Su, W., Cao, R., Fujita, M., Zhou, Z. & Chan, A. S. C. (2000*a*). *Angew. Chem. Int. Ed.* **39**, 2468–2470.
- Hong, M., Zhao, Y., Su, W., Cao, R., Fujita, M., Zhou, Z. & Chan, A. S. C. (2000*b*). *J. Am. Chem. Soc.* **122**, 4819–4820.
- Janiak, C. (2000). *J. Chem. Soc. Dalton Trans.* pp. 3885–3896.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.